# Basic Water and Wastewater Formulas 

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## Summary

Operators obtaining or maintaining their certification must be able to calculate complex formulas and conversion factors. This Tech Brief provides basic examples of these formulas and conversion factors.

## Metric Conversion Factors (Approximate) Conversions from Metric Measures

| Symbol | Whe | \# You Know | Mortiply By | To Find | Symbol |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Length Area | mm | millimeters | 0.04 | inches | in |
|  | cm | centimeters | 0.4 | inches | in |
|  | m | meters | 3.3 | feet | ft |
|  | m | meters | 1.1 | yards | yds |
|  | km | kilometers | 0.6 | miles | mi |
| Mass (Weight) | $\mathrm{cm}^{2}$ | square centimeters | 0.16 | square inches <br> square yards <br> square miles <br> acres | $\begin{aligned} & \mathrm{in}^{2} \\ & \mathrm{yd}^{2} \\ & \mathrm{mi}^{2} \\ & \mathrm{acrs} \end{aligned}$ |
|  | $\mathrm{m}^{2}$ | square meters | 1.2 |  |  |
|  | $\mathrm{km}^{2}$ | square kilometers | 0.4 |  |  |
|  | ha | hectares ( $10,000 \mathrm{~m}^{2}$ ) | 2.5 |  |  |
| Volume Temp ${ }_{\mathrm{k}}^{\mathrm{g}}$$F=(9 / 5) C+32 t$ |  | grams | 0.035 | ounces <br> pounds <br> short tons | $\begin{aligned} & \text { oz } \\ & \text { lbs } \end{aligned}$ |
|  |  | kilograms tones ( $1,000 \mathrm{~kg}$ ) | 2.2 |  |  |
|  |  | 1.1 |  |  |
|  | ml |  | milliliters | 0.03 | fluid ounces <br> pints <br> quarts <br> gallons <br> cubic feet <br> cubic yards | $\begin{aligned} & \mathrm{fl} \mathrm{oz} \\ & \mathrm{pt} \\ & \mathrm{qt} \\ & \mathrm{gal} \\ & \mathrm{ft}^{3} \\ & \mathrm{yd}^{3} \end{aligned}$ |
|  | 1 | liters | 2.1 |  |  |  |
|  | 1 | liters | 1.06 |  |  |  |
|  | 1 | liters | 0.26 |  |  |  |
|  | $\mathrm{m}^{3}$ | cubic meters | 35.0 |  |  |  |
|  | $\mathrm{m}^{3}$ | cubic meters | 1.3 |  |  |  |
|  | ${ }^{0} \mathrm{C}$ | Celsius temperature | $\begin{aligned} & 9 / 5 \text { (then } \\ & \text { add } 32 \text { ) } \end{aligned}$ | Fahrenheit temperature | ${ }^{0} \mathrm{~F}$ |  |



## Basic Water and Wastewater Formulas

Area, $f t^{2} \quad$ Rectangle, Width, $f t \mathrm{x}$ Length, $f t$
Circle, $\quad(\text { Diameter, } f t)^{2} \pi 4$
Backwash Rate, $g p m / f t^{2} \quad$ Flow, $g p m$
Area, $f t^{2}$
Filtration Rate, $g p m / f t^{2}$
Flow, gpm
Area, $f t^{2}$
Chlorine Dose, $m g / L$
$\mathrm{Cl}_{2}$ Demand, $m g / L+$ Free $\mathrm{Cl}_{2}$ Residual, $m g / L$

Circumference
of a circle, $f t$
$(\pi)$ (Diameter, $f t$ )
or
$2(\pi)$ (Radius, $f t$ )
Detention time, hrs (Volume, gal)(24 hrs/day) Flow, gpd

Flow, cfs
(Velocity, $f t / s e c$ ) (Area, $f t^{2}$ )

Velocity, ft/sec
Flow, cfs
Area, $f t^{2}$

Distance,ft.
Time, sec.

Water Horsepower, HP (Flow, gpm) (Head, ft) 3960

Pounds, lbs (Flow, MGD)(Conc. $m g / L$ )(8.34 lbs/gal)

Power, watts
(Voltage, volts)(Current, amp)
Power Factor Actual Power, watts Apparent Power, V-A

Removal, \%
In - Out In
Solution Strength, \% Weight of Chemical Weight of Solution $\times 100$

Surface Overflow Rate, gpd/ft ${ }^{2}$ Flow, gpd Area, $\mathrm{ft}^{2}$

| Temperature | ${ }^{\circ} \mathrm{F}\left(1.8 \mathrm{x}{ }^{\circ} \mathrm{C}\right)+32$ |
| :--- | :---: |
|  | ${ }^{\circ} \mathrm{C}\left({ }^{\circ} \mathrm{F}-32\right)(5 / 9)$ |
| Velocity, $f t / \mathrm{sec}$ | Flow, $f t^{3} / \mathrm{sec}$ |
|  | _ Area, $f t^{2}$ |

Volume, $f t^{3}$
Rectangle; Width, ft $\times$ Length, ft $\times$ Height, ft
Cylinder; $\frac{\pi \text { (Diameter, } f t)^{2}}{4} \underline{(\text { Height, } \mathrm{ft})}$
Cone; $\boldsymbol{\pi}$ (Diameter, ft$)^{2}(\underline{\text { Height,ft) }} 12$
Sphere; $\pi$ (Diameter, ft$)^{3}$


Reservoir Volume, gal. =
Volume, ac-ft x $43,560 \mathrm{ft}^{2} / \mathrm{ac} . \mathrm{x} 7.48 \mathrm{gal} / \mathrm{ft}^{3}$
Reservoir Surface Area, ac. =

$$
\frac{\text { Surface Area, } \mathrm{ft}^{2}}{43,560 \mathrm{ft}^{2} / \mathrm{ac}}
$$

Slope $=\quad \underset{\text { Length, } \mathrm{ft}}{\text { Fall, } \mathrm{ft}}$
Grade $=\quad \frac{\text { Rise, } \mathrm{ft}}{\text { Run, } \mathrm{ft}}$

## Conversion Factors

| $1 \mathrm{ft}^{3}$ water $=7.48 \mathrm{gal}$ | 1 liter $/ \mathrm{sec}=15.85 \mathrm{gpm}$ | 1 kilowatt $=1.34 \mathrm{HP}$ |
| :--- | :--- | :--- |
| $1 \mathrm{yd}^{3}=27 \mathrm{ft}^{3}$ | $1 \mathrm{acre}=43,560 \mathrm{ft}^{2}$ | $1 \mathrm{HP}=550 \mathrm{ft}-\mathrm{lbs} / \mathrm{sec}$ |
| 1 gal water $=8.34 \mathrm{lbs}$ | $1 \mathrm{psi}=2.31$ feet of water | $1 \mathrm{HP}=0.746$ kilowatts |
| $1 \mathrm{ft}^{3}$ water $=62.4 \mathrm{lbs}$ | $1 \mathrm{mg} / \mathrm{L}=1 \mathrm{ppm}$ | 1 meter $=3.28$ feet |
| $1 \mathrm{MGD}^{*}=694 \mathrm{gpm}$ | $1 \%=10,000 \mathrm{mg} / \mathrm{L}$ | 1 mile $=5280$ feet |
| $1 \mathrm{MGD}=1.547 \mathrm{cfs}$ | 1 kilogram $=2.20 \mathrm{lbs}$ | 1 kilopascal $=0.145 \mathrm{psi}$ |
| 1 liter $=0.264 \mathrm{gal}$ | 1 centimeter $=0.394$ inches | $\pi(\mathrm{Pi})=3.1416$ |

*MGD = million gallons per day

## Powers of Ten

Prefixes and symbols to form decimal multiples and/or submultiples.

| Power <br> of Ten | E <br> Notation | Decimal <br> Equivalent | Prefix | Phonic | Symbol |
| :--- | :--- | :--- | :--- | :--- | :--- |
| $10^{12}$ | $\mathrm{E}+12$ | $1,000,000,000,000$ | tera | ter'a | T |
| $10^{9}$ | $\mathrm{E}+09$ | $1,000,000,000$ | giga | ji'ga | G |
| $10^{6}$ | $\mathrm{E}+06$ | $1,000,000$ | mega | meg'a | M |
| $10^{3}$ | $\mathrm{E}+03$ | 1,000 | kilo | kil'o | k |
| $10^{2}$ | $\mathrm{E}+02$ | 100 | hecto | hek'to | h |
| 10 | $\mathrm{E}+01$ | 10 | deka | dek'a | da |
| $10^{-1}$ | $\mathrm{E}-01$ | 0.1 | deci | des'I | d |
| $10^{-2}$ | $\mathrm{E}-02$ | 0.01 | centi | sen'ti | c |
| $10^{-3}$ | $\mathrm{E}-03$ | 0.001 | milli | mil'I | m |
| $10^{-6}$ | $\mathrm{E}-06$ | $0.000,001$ | micro | mi'kro | u |
| $10^{-9}$ | $\mathrm{E}-09$ | $0.000,000,001$ | nano | nan'o | n |
| $10^{-12}$ | $\mathrm{E}-12$ | $0.000,000,000,001$ | pico | pe'ko | p |
| $10^{-15}$ | $\mathrm{E}-15$ | $0.000,000,000,000,001$ | femto | fem'to | f |
| $10^{-18}$ | $\mathrm{E}-18$ | $0.000,000,000,000,000,001$ | atto | at'to | a |

## Sample Questions

1. An empty storage tank at standard atmospheric pressure (not under pressurized condition) is 8 feet in diameter and 32 feet high. How long will it take to fill 90 percent of the tank volume if a pump is discharging a constant 24 gallons per minute into the tank?
a. 7 hours and 31 minutes
b. 8 hours and 21 minutes
c. 8 hours and 23 minutes
d. 9 hours and 17 minutes

Solution: Don't look at the problem as a whole. Instead, break it into steps:

First, calculate the area of a circle 8 feet in diameter;
$\frac{(\text { Diameter, } \mathrm{ft})^{2}}{4} \pi, \frac{8}{4}^{2} \pi, \frac{64}{4} \pi, 16(3.1416)=50.26 \mathrm{ft}^{2}$
Second, calculate the volume of a cylinder;
$(\text { Diameter, } \mathrm{ft})^{2} \pi$ (Height, ft),
4 since the area is already calculated, just multiply by the height.
$50.26 \mathrm{ft} 2 \times 32 \mathrm{ft}$ high $=1,608.5 \mathrm{ft}^{3}$ (cubic feet)
Third, convert from $\mathrm{ft}^{3}$ (cubic feet) to gallons; $1 \mathrm{ft}^{3}$ water $=7.48$ gallons


Fourth, calculate what $90 \%$ of the total volume would be
$12,031.5$ gallons $\times \frac{90 \%}{100}$,
$12,031.50$ gallons $\mathrm{x} .90=10,828.39$ gallons
Fifth, calculate time to pump at 24 gallons per minute

10,828.39 gallons $=451.18$ minutes 24 gallons minute

Sixth, convert minutes to hours and minutes; 60 minutes $=1$ hour
451.18 minutes $=7.52$ hours 60 minutes
hour
Now take the .52 hours and multiply by 60minutes/hour

7 hours and $(0.52 \times 60)=31.2$ minutes
7 hours and 31 minutes
The answer is $a .7$ hours and 31 minutes

## 2. How many cubic feet of water will a rectangular tank that is 20 -feet long by 15 -feet wide and 10 -feet high hold? <br> a,000 cubic feet <br> b. 3,000 cubic feet <br> c. 850 cubic feet <br> d. 1,200 cubic feet

## Solution:

Calculate the volume for a rectangular box ( $\mathrm{L} \times$ W x D or H) Length x Width x Depth or Height.
$20 \mathrm{ft} \times 15 \mathrm{ft} \times 10 \mathrm{ft}=3,000 \mathrm{ft}^{3}$ (cubic feet)
The answer is b. 3,000 cubic feet

## 3. Calculate the chlorine demand using the following data:

Raw water flow is 0.75 MGD
Chlorinator feed rate is $4.0 \mathrm{mg} / \mathrm{L}$
Chlorine residual (free) is $1.8 \mathrm{mg} / \mathrm{L}$
a. $\quad 0.8 \mathrm{mg} / \mathrm{L}$
b. $2.2 \mathrm{mg} / \mathrm{L}$
c. $4.0 \mathrm{mg} / \mathrm{L}$
d. $5.8 \mathrm{mg} / \mathrm{L}$

## Solution:

If solution strength is not given, then use $100 \%$
Often more information is given than needed to solve specific problems. In this problem, the raw water flow rate ( 0.75 MGD ) is not needed.

The equation to be used is the Chlorine Dose Equation

Chlorine Dose, mg/L =
$\left(\mathrm{Cl}_{2}\right.$ Demand, $\left.\mathrm{mg} / \mathrm{L}\right)+$ Free $\mathrm{Cl}_{2}$ Residual, $\mathrm{mg} / \mathrm{L}$
Solve this Equation for the Chlorine Demand $\mathrm{Cl}_{2}, \mathrm{mg} / \mathrm{L}$;
$\mathrm{Cl}_{2}$ Demand, mg/L=
Chlorine Dose, mg/L - Free $\mathrm{Cl}_{2}$ Residual, mg/L
$\mathrm{Cl}_{2}$ Chlorine Demand, $\mathrm{mg} / \mathrm{L}=$
$4.0 \mathrm{mg} / \mathrm{L}-1.8 \mathrm{mg} / \mathrm{L}$
$\mathrm{Cl}_{2}$ Chlorine Demand, $\mathrm{mg} / \mathrm{L}=2.2 \mathrm{mg} / \mathrm{L}$
The answer is b. $2.2 \mathrm{mg} / \mathrm{L}$
4. Calculate the volume, in gallons, of a tank that is $\mathbf{7 5}$ feet long, 20 feet high, and 10 feet deep.
a. 15,000 gallons
b. 112,200 gallons
c. 150,000 gallons
d. 224,400 gallons

Solution:
First, calculate the volume for a rectangular box (L x W x D or H) Length x Width x Depth or
$75 \mathrm{ft} \times 20 \mathrm{ft} \times 10 \mathrm{ft}=15,000 \mathrm{ft}^{3}$ (cubic feet)
Second, convert $\mathrm{ft}^{3}$ (cubic feet) to gallons, $1 \mathrm{ft}^{3}$ (cubic feet) water $=7.48$ gallons

$$
\begin{aligned}
15,000 \mathrm{ft}^{3} \times \underset{\mathrm{ft}^{3}}{7.48 \text { gallons }}= & \begin{array}{l}
112,200 \text { gallons, } \\
\text { this is the tank } \\
\text { volume in gallons }
\end{array}
\end{aligned}
$$

The answer is b. 112,200 gallons
5. How many pounds of a chemical applied at the rate of $3 \mathrm{mg} / \mathrm{L}$ are required to dose $\mathbf{2 0 0}, 000$ gallons?
a. 1 lb
b. 3 lbs
c. 5 lbs
d. 16 lbs

## Solution:

If a solution strength is not given, use $100 \%$
The equation to be used is the pounds, lbs equation - (Flow, MGD)(Conc.mg/L)(8.34 lbs/gal)

Convert the flow or gallons to MGD;
$\xlongequal{200,000}$ gallons per day $=0.2 \mathrm{MGD}$
1,000,000 million gallons
The concentration is the rate in this case $=3 \mathrm{mg} / \mathrm{L}$
Now plug the givens (known) into the pounds equation
$(0.2, \mathrm{MGD})(3 \mathrm{mg} / \mathrm{L})(8.34 \mathrm{lbs} /$ gal, day $)=5.004 \mathrm{lbs}$
The answer is c. 5 lbs

## References:

Glover, Thomas J. 1999. Pocket Reference. Sequoia Publishing, Inc.: Littleton, CO.
National Environmental Services Center, 2007. Basic Water \& Wastewater Formulas Product \# DWPCOM84.
U.S. Navy. Physics Formulas notebook sheet.

Virginia Center for Very Small Water Works. 2003. Sample Test Questions. http:/ / www.vaclassix.com/files/VSWS_samplequestions_answers.pdf


